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Chiral Phosphoric Acid Catalysis Mechanism



BINOL-Derived Chiral Phosphoric Acids

- Brønsted acids have proven to be highly efficient and versatile catalysts.
- Most prominent players are BINOL (1,1'-bi-2-naphthol) derived phosphoric acids (CPAs).
- First described independently by Terada and Akiyama groups in 2004.^[1,2]

Previous Mechanistic Studies

• Transfer hydrogenation of imines with Hantzsch ester was established as a model system for mechanistic investigations:[3]



- Besides experimental and spectroscopic investigations, computational studies have been key to gaining deeper understanding of the reactivity and stereoselectivity of CPAcatalysed reactions.
- In 2016 Goodman et al. linked the stereoselectivity to two parameters reflecting the steric properties of the 3,3'-substituents and developed the 'Goldilocks' model for predicting suitable CPAs based on reactant structures.^[4]
- Independent studies by Sigman et al. established a data-driven prediction model to elucidate the influence of multiple parameters of the reagents on the stereoselectivity.^[5]





A simple and straightforward prediction of the optimal catalyst for each substrate/reactant combination remained extremely difficult.

References: [1] Angew. Chem. Int. ed. 2004, 43, 27, 1566. [2] J. Am. Chem. Soc. 2004, 126, 5356. [3] J. Am. Chem. Soc. 2008, 130, 8741. [4] J. Am. Chem. Soc. 2016, 138, 7910. [5] Nature, 2019, 571, 343.

Controlling the Competition between Monomeric versus Dimeric Reaction Pathways Enhances Stereoselectivities^[6]

- Niemeyer et al. already showed in 2020 that in addition to the generally postulated monomeric pathway, a dimeric pathway with an opposite enantioselectivity exists.^[7]
- By separating the monomeric and dimeric pathways, significantly improved stereoselectivities were achieved for both product enantiomers by Gschwind et al.^[6]
- Evidence from NMR studies, DFT computations and catalytic reactions.

Enantioselectivity is dependent on temperature, concentration and catalyst loading

Hypothesis

- Explanation for high previous reactions temperatures
- The kinetic preference of the dimeric pathway is almost independent of the substrate
- The dimeric pathway could be influential in various CPAcatalysed reactions beyond just the transfer hydrogenation of quinolines or imines
- The provided guideline regarding temperature and catalyst proportion to control the competition between monomeric and dimeric pathway can be transferred as a general scheme



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References: [6] Angew. Chem. Int. Ed. 2023, 62, e202301183. [7] Chem. Sci. 2020, 11, 4381.



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